Corrections to Answers in Tro (4th Ed)

The following list corrections is likely not complete. Please let us know if you find others.

Chapter 2

#53 All neutrons should have the symbol

Chapter 3

• #145 Significant figures make a big difference in this answer. Answer may range from 0.210 to 0.225.

 $\frac{1}{n}$

Chapter 4

• #47b 2.91 g of CO (not CO₂) remain.

Chapter 5

- #79 4.84 g
- #137 No units (Atm should not be in the answer.)

Chapter 6

#83 $1/2 \text{ N}_2(g) + 3/2 \text{ H}_2(g) \rightarrow \text{NH}_3(g)$

Chapter 9

- #47 -717 kJ/mole
- #63a Structure shown below is preferred by the rule of minimizing formal charge.
- #75d Structure shown below is preferred by the rule of minimizing formal charge.

••	••	••		••	••	••
· O_	-CI-	-0	<>	0-	- CI_	<u>_</u> O·
	0	$-\circ$				
••	••	••		••	••	••
-1	0	0		0	0	-1

• #83 Answer is OK because there is an erro in the equation given in the textbook. The correctly balanced equation has 2 moles of H₂ in the products.

 $C(s) + 2 H_2O(g) \longrightarrow \underline{2} H_2(g) + CO_2(g)$

 #107b H₃PO₃ is an exception to the rules of oxoacids where H is usually attached to an oxygen atom and not the central atom. The structure shown is the experimentally verified structure.

Chapter 10

- #51 c BrCl₅ is polar
- #97b Omit this problem. There is not enough information to predict this structure.